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compound is 55.3% k, 14.6% p and 30.1% o what is the empirical formula assume 100 g of compound we have 55.3 g of k, 14.6 g p and 30.1 g o look up the molar masses of k, p and o and convert these to moles

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web strategy 4 be careful with negatives adding negatives such as not do not necessarily make a statement false hint re phrase the sentence to help you clarify what is being asked or stated if the negative words confuse you example 1 california is not a state in canada re phrase canada does not have a state called california or california is not in canada

web may 28 2020 determine the empirical formulas for compounds with the following percent compositions a 15.8% carbon and 84.2% sulfur b 40.0% carbon, 6.7% hydrogen and 53.3% oxygen answer a CS<sub>2</sub> answer b CH<sub>2</sub>O click here to see a video of the solution

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